

Learning Task Inventories (LTIs) in Introductory Organic Chemistry

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The Western Conference on Science Education



ORGANIC CHEMISTRY I AT WLU

- first organic chemistry course
- required for many Year 2 Science students
- prior to 2010 → primarily lecture based
- starting in 2010 → "blended" approach
- course enrollment



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WHY LTIs?

blended learning: start with learning objectives

Your ability to predict chemical reactivity of organic compounds and illustrate mechanisms by which reactions occur is highly dependent on your ability to *visualize* and *represent* the chemical structures of organic compounds and to *explain* the effect that structure has on reactivity. By the end of this course, you should be able to:

- ◆ describe and illustrate the structure of atoms and the nature of bonding between atoms in organic compounds;
- ◆ evaluate the effect that bonding has on the 3-D shape, stability, properties and reactivity of organic compounds;
- ◆ assess the consequences of 3-D shape, conformational mobility and stereochemistry on structure and reactivity of organic compounds;
- ◆ provide the chemical structures of reactants, reagents or products required for or predicted by a given chemical reaction:
- solve and/or depict, using the curved arrow formalism (a.k.a. arrow pushing), the mechanisms by which organic reactions occur.
- "What do we have to know...?"
- develop self-monitoring, metacognitive skills

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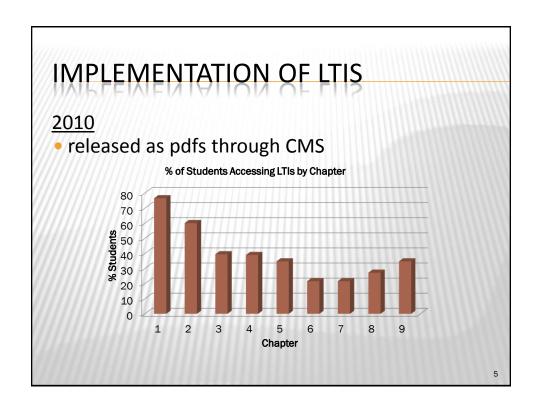
TYPICAL LEARNING TASK INVENTORY

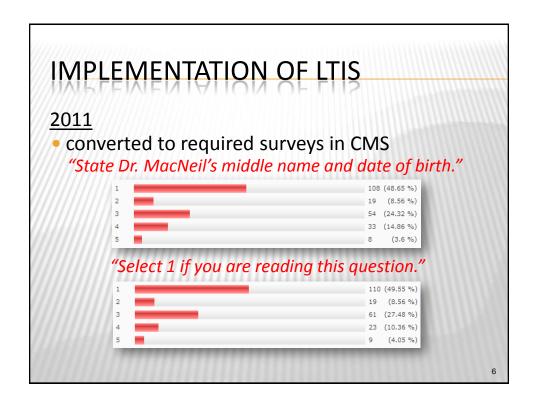
Learning Tasks for Chapter 6. An Overview of Organic Reactions

After covering the material from this chapter, you should be able to:

- ♦ name, describe and recognize examples of the four common types of organic reactions
- ◆ distinguish between symmetrical bond breaking/making (radical) mechanisms and unsymmetrical bond breaking/making (polar) mechanisms, and dilustrate how each type occurs using curved arrows
- ◆ name, describe and illustrate, using curved arrows, the three mechanistic steps involved in the radical halogenation of alkanes
- ◆ define nucleophile and electrophile, identify molecules, and atoms within molecules, that are capable of acting as either or both, and use curved arrows to show the movement of electron density when nucleophiles and electrophiles react (IMPORTANT AND ON-GOING TASKS IN THIS COURSE FROM THIS POINT ON)
- ullet define ΔG° and $\Delta G^{\bar{1}}$ for a given reaction, or for a particular mechanistic step in a reaction, and sketch an energy diagram given information regarding the sign and magnitude of ΔG° and $\Delta G^{\bar{1}}$
- ullet describe the relationship between ΔG° and K_{eq} , and between ΔG^{\ddagger} and the rate of a particular step in a reaction
- ◆ define exergonic and endergonic, and, given an energy diagram_classify_a reaction, or an individual mechanistic step of a reaction, as exergonic or endergonic
- given an energy diagram, <u>locate</u> starting materials, transition states, intermediates and products, and <u>identify</u> the rate-limiting step
- ♦ for the acid-catalyzed addition of water to ethylene, <u>draw</u> a detailed arrow-pushing mechanism and an energy diagram labeled with the rate-limiting step and the locations and structures of transition states and intermediates

◆ *state* the differences between common laboratory reactions and biological reactions





LTI RESEARCH DESIGN

2012

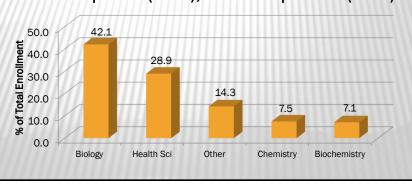
- 293 students (94%) in 'Orgo 1' recruited
- randomly divided into 5 treatment groups
- completed introductory and end-of-term surveys and 9 weekly LTIs

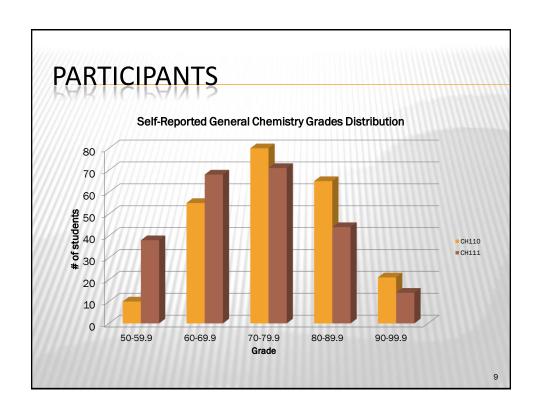
Condition	1	2	3	4	5
LTI	V	1	$\sqrt{}$	\checkmark	\checkmark
Prompt	Х	$\sqrt{}$	V	\checkmark	$\sqrt{}$
Quiz	X	X	(no feedback)	(part. feedback)	(full feedback)
Survey	Х	Χ	$\sqrt{}$	$\sqrt{}$	$\sqrt{}$

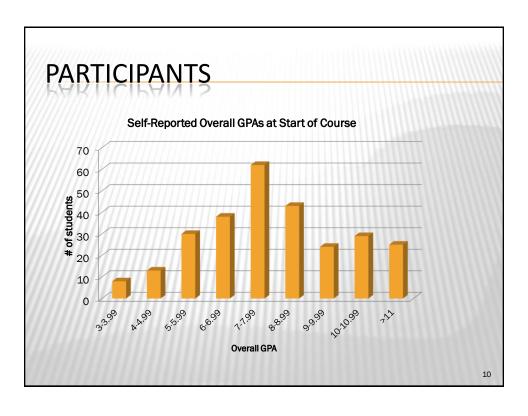
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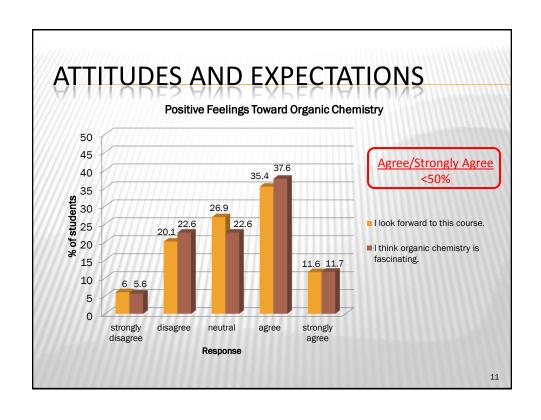
PARTICIPANTS

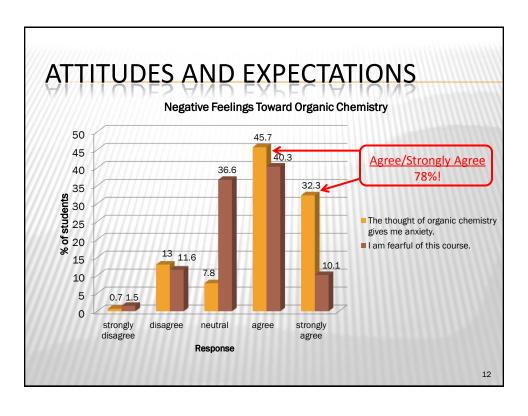
- 293 participants
- 186 females (69%); 82 males (31%)
- 233 2nd year (86%); 37 3rd, 4th year (14%)
- course required (87%); course optional (13%)

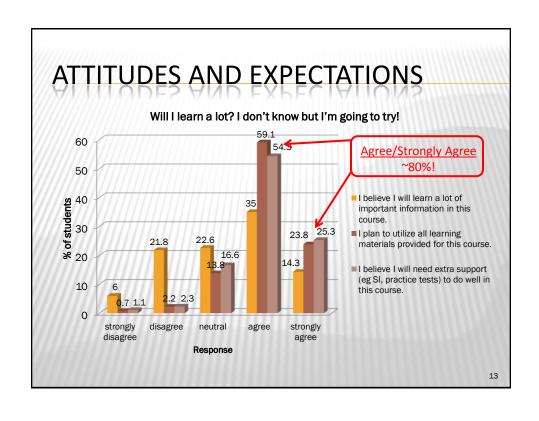




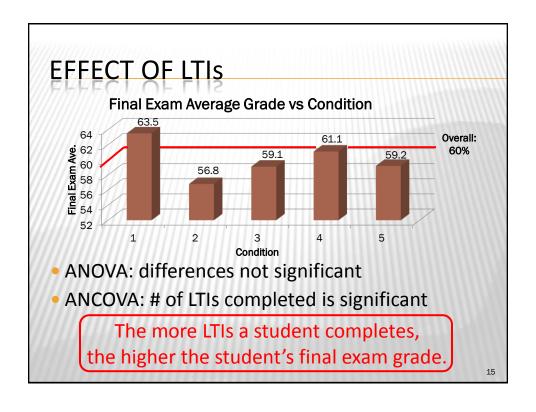


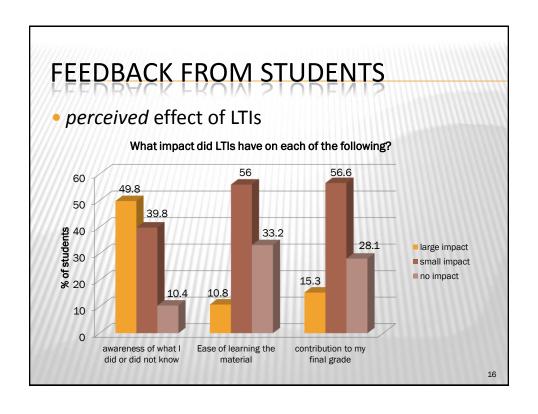


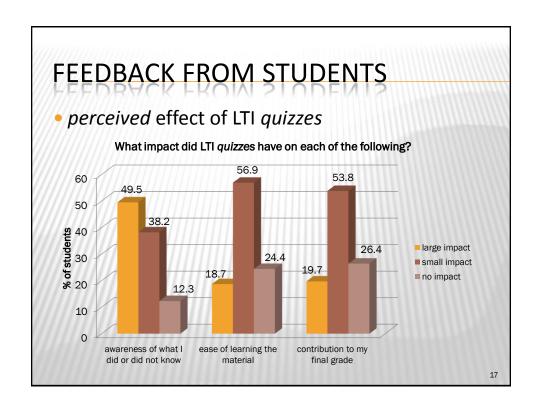


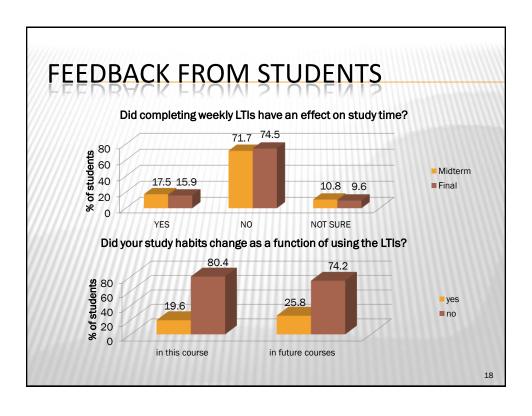


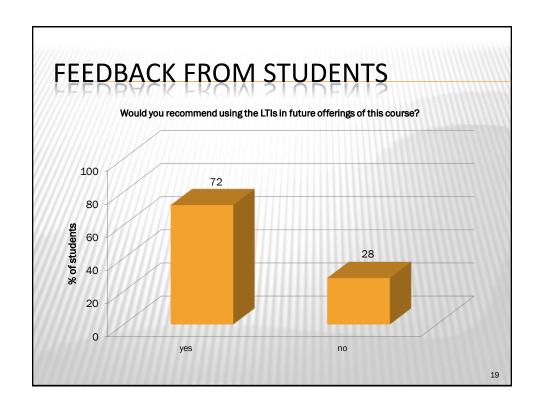


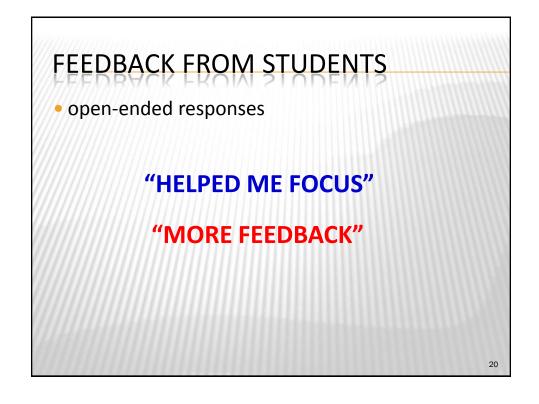












CONCLUSIONS

- LTI conditions did not have a significant effect on final exam grades
- # of LTIs completed was a significant predictor of final exam grades even after controlling for prior learning
- students feel that LTIs improve "awareness" or "focus" but do not think this translates to improved study habits or grades

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FUTURE WORK

- directly measure metacognitive skills at beginning and end of course
- explore effect of LTI frequency on improvement of metacognitive skills
- hold interviews to gain insight into how students are using the LTIs
- develop scaffolding that supports other aspects of self-regulated learning

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